

## Polishing characteristics of single crystal SiC assisted by plasma oxidation using different kinds of abrasives

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### Abstract

For the flattening of some difficult-to-machine materials, such as 4H-SiC, GaN, diamond and so on, we proposed a precision polishing technique named plasma assisted polishing (PAP), which combined the irradiation of atmospheric pressure water vapor plasma and soft abrasive polishing. For the optimization of the abrasive material in PAP, the polishing characteristics of different abrasives such as CeO<sub>2</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and diamond are evaluated in this work. In the case of polishing using CeO<sub>2</sub> and SiO<sub>2</sub> slurry, step-terrace structure of SiC was generated and that on the CeO<sub>2</sub> slurry polished surface was much more uniform. In contrast, scratches were introduced on the surfaces polished by Al<sub>2</sub>O<sub>3</sub> and diamond slurry due to the high hardness of the abrasive materials.

### 1. Introduction

4H-SiC is considered one of the most promising next-generation semiconductor materials owing to its excellent mechanical and electrical properties, such as high hardness, wide band gap, and high thermal conductivity, and so on. To make these properties to sufficiently good use, an atomically flat and damage-free surface is essential. We proposed a novel polishing technique named PAP, which combines modification by atmospheric pressure water vapor plasma and polishing using soft abrasive [1, 2]. Although the good smoothing ability of PAP was already proved, the material removal rate (MRR) of PAP was very low. In order to increase the MRR of PAP, optimization of the modification process and polishing process in PAP is essential. In this paper, the polishing characteristics of 4H-SiC (0001) using different abrasives such as CeO<sub>2</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and diamond are evaluated.

## 2. Experimental setup and parameters

Fig. 1 shows the experimental setup for water vapor plasma oxidation. The 4H-SiC specimen was set on the ground electrode. Helium-based water vapor (1.5 standard liters per minute) was used as the process gas and was introduced into the space between the electrodes. The diameter of the aluminum alloy powered electrode was 4 mm. Atmospheric-pressure helium-based water vapor plasma was generated by applying a 13.56 MHz RF power between the electrodes. The RF power was 45 W with a duty ratio of 10%. The gap between the powered electrode and the SiC wafer was 1 mm. The polishing process was conducted using a slurry polishing system. The specimen was immersed into slurry and the polishing pad scanned on it. Table 1 shows the conditions of slurry polishing.

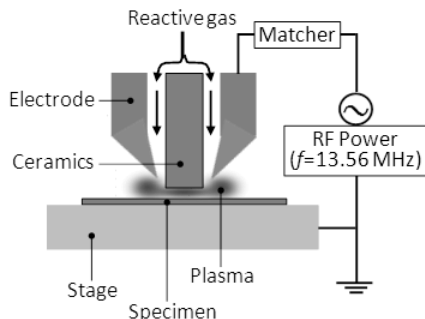


Figure 1: Experimental setup of water vapor plasma irradiation.

Table1: Experimental conditions of slurry polishing.

Parameters	Conditions
Load	50 g
Polishing pad	K0017 (FILWEL Co. Ltd.) ( $\phi=10$ mm)
Abrasive material	CeO <sub>2</sub> ( $\phi=190$ nm), SiO <sub>2</sub> ( $\phi=80$ nm), Al <sub>2</sub> O <sub>3</sub> ( $\phi=100$ nm), diamond ( $\phi=100$ nm)
Polishing time	3 h
Rotation speed	2000 rpm
Slurry concentration	CeO <sub>2</sub> (1.0 wt%), SiO <sub>2</sub> (1.0 wt%), Al <sub>2</sub> O <sub>3</sub> (10.0 wt%), diamond (0.4 wt%)

## 3. Results and discussion

Figure 2 shows an atomic force microscopy (AFM) image of a SiC (0001) surface processed by plasma oxidation followed by slurry polishing. An atomically flat surface with a well-ordered step-terrace structure is obtained. This result indicates

that combination of plasma oxidation and CeO<sub>2</sub> abrasive polishing is a very useful method for surface flattening of 4H-SiC.

In a PAP process, when an oxide layer is generated, it will be immediately removed by abrasive polishing. After the oxide layer was removed, polishing of the bulk SiC will be conducted. Therefore, the surface morphology obtained by PAP was determined by the oxide-SiC interface

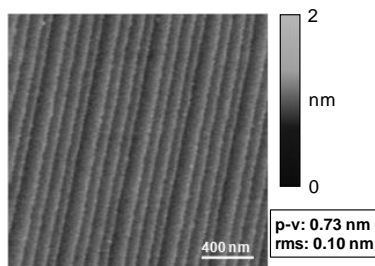


Figure 2: AFM image of a SiC (0001) surface processed by plasma oxidation followed by slurry polishing.

and the abrasive polishing characteristic of 4H-SiC. Therefore, polishing of 4H-SiC (0001) using different abrasives such as CeO<sub>2</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and diamond were respectively conducted without previous plasma oxidation.

Figure 3 shows the AFM images of SiC (0001) surfaces processed by polishing using different abrasives. When CeO<sub>2</sub> and SiO<sub>2</sub> slurry was used, step-terrace structure of SiC was generated. Although the hardness of CeO<sub>2</sub> and SiO<sub>2</sub> is much lower than that of SiC, SiC is oxidized by chemical reactions in slurry and the oxide layer can be removed by CeO<sub>2</sub> and SiO<sub>2</sub> abrasives. The interaction between chemical reaction and mechanical polishing greatly affect the morphology of the step-terrace structure. Because the oxidation rates of the four SiC bilayers in one unit cell of 4H-SiC are different whereas the mechanical removal rates of these bilayer are supposed to be the same. In the case of CeO<sub>2</sub> slurry polishing, well-ordered step-terrace structure is observed, thus, the mechanical removal rate should be higher than the oxidation rate in this process because the MRR of SiO<sub>2</sub> by CeO<sub>2</sub> slurry polishing is high. In contrast, when SiO<sub>2</sub> slurry was used, a step-terrace structure with alternations of four kinds of terraces is observed. It means that the oxidation rate in this process is higher than the mechanical removal rate. And this is why a periodic step-terrace structure is generated. It can be found that the terrace-edge is not uniform and it is considered that a very small amount of oxide exist at the terrace-edge.

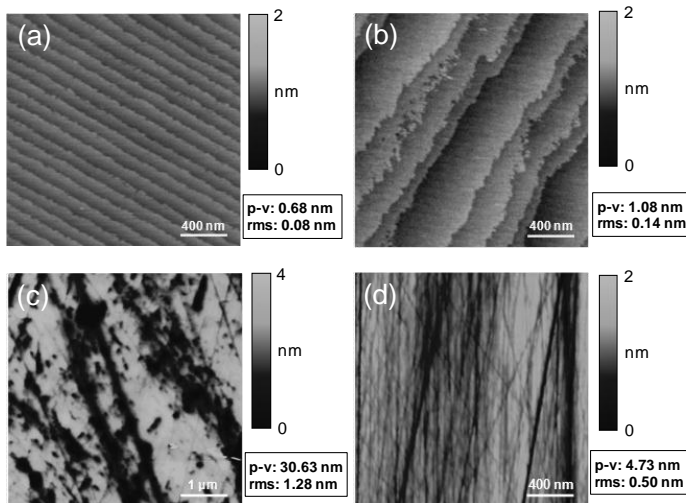


Figure 3: AFM image of a SiC (0001) surface processed by polishing using different abrasives. (a) CeO<sub>2</sub>, (b) SiO<sub>2</sub>, (c) Al<sub>2</sub>O<sub>3</sub>, (d) diamond.

When Al<sub>2</sub>O<sub>3</sub> and diamond slurry are applied, many scratches are introduced as shown in Figure 3 (c) and (d). This is due to the high hardness of Al<sub>2</sub>O<sub>3</sub> and diamond

#### 4. Conclusions

The polishing characteristics of CeO<sub>2</sub>, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and diamond were evaluated in this study. No scratches were introduced when CeO<sub>2</sub> and SiO<sub>2</sub> were used. CeO<sub>2</sub> was very useful for the generation of a well-ordered step-terrace structure.

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